

Lattice Energy

The energy change when 1 mole of a solid compound is formed from its gaseous ions under standard conditions.

Entropy of a system

The measure of the disorder of a system

Standard enthalpy change of solution

The energy change when 1 mole of substance dissolves in water

Standard enthalpy change of hydration

Energy change when 1 mole of gaseous ions dissolve in water to form an infinitely dilute solution

Electron Affinity

The enthalpy change when 1 mole of electrons is gained by 1 mole of gaseous ions



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Not published yet.

Last updated 20th November, 2024.

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