

Lattice Energy

The energy change when 1 mole of a solid compound is formed from its gaseous ions under standard conditions.

Entropy of a system

The measure of the disorder of a system

Standard enthalpy change of solution

The energy change when 1 mole of substance dissolves in water

Standard enthalpy change of hydration

Energy change when 1 mole of gaseous ions dissolve in water to form an infinitely dilute solution

Electron Affinity

The enthalpy change when 1 mole of electrons is gained by 1 mole of gaseous ions



By [youngsdoubleslit](https://cheatography.com/youngsdoubleslit/)

cheatography.com/youngsdoubleslit/

Not published yet.

Last updated 20th November, 2024.

Page 1 of 1.

Sponsored by [ApolloPad.com](https://apollopad.com)

Everyone has a novel in them. Finish

Yours!

<https://apollopad.com>