

Reversible and Irreversible reactions

Reversible reaction: A reaction in which not only the reactant react to form product but also the products react to form reactants under the same conditions is called reversible reaction. In general, a reaction that not only takes place in the forward direction but, also in the reverse direction.

Irreversible reactions: A reaction which cannot take place in reverse direction is called irreversible reaction.

Equilibrium

A reaction is in equilibrium when the rate of forward reaction is equal to the rate of backward reaction. Only the reversible reaction can attain equilibrium.

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By **NIDHIYA (NIDHIYA)**
cheatography.com/nidhiya/

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Page 1 of 1.

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