

Biochemistry Lecture 1 Cheat Sheet by Morghay123 via cheatography.com/53154/cs/14366/

Precursors	
Linear Polymers	- Covalent bonds - nucleotides in DNA and RNA - too big to fit in a cell - fold into complex 3D shapes
Supramolecular Complexes	- Groups of linear polymers -enzymes, ribosomes
Subcellular Organelles	- Golgi - ER - Group of supramolecular complexes
> Cells> Tissues	> Organisms

	complexes	
> Cells>	> Organisms	
Tissues		
Cell Requirements to live, grow and propagate		
Precursors (that which comes before)	precursors are oxidized for energy or combined to form larger biomolecules	
Energy	to drive chemical reactions and/or other cellular processes (ex: muscle contractions)	
Information	to direct and control cellular growth, propagation and response to environmental change	

Stabilizing Forces	
London Forces (dispersio n, induced dipoles)	- Weakest - e;ectron clouds overlap, short lifespan, constantly breaking and reforming
Hydropho bic Interaction s	- Special london force, non-polar molecules dissolved in water. the two don't mix so H2O forms a rigid structure around the non-polar molecules
Dipole-Dipo	le Interactions
Hydrogen Bond	- Strong Dipole-dipole interaction - attached to N, O, F (F not in living cells) -lots of attractive force
Electrosta tic Interaction s	lon pairing or salt bridges
molecules in	rease = more stable when nonpolar nteract with each other surrounded
General De	finitions
Biochemistry: the study of the matter that	

change tog	together	
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	molecules	
Dipole-Dipole Interactions		
Hydrogen Bond	- Strong Dipole-dipole interaction - attached to N, O, F (F not in living cells) -lots of attractive force	
Electrosta tic Interaction s	lon pairing or salt bridges	
Energy decrease = more stable when nonpolar molecules interact with each other surrounded by water - The 5 hold life together		
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General De	etinitions	
Biochemistry: the study of the matter that makes up living things		
Structural Complementarity: what holds us together		
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Classes of Precursor Molecules		
Carbohydr ates	PRIMARY ENERGY SOURCE -Most abundant molecule in biosphere -Sugars -Play a role in other molecules	
Lipids (fats and oils)	- excess of everything -energy reserve (resevoir) -can store an infinite amount - stored in adipose cells - MAIN COMPONENTS OF THE CELL MEMBRANE (barrier)	
Amino Acids	- serve as a precurser to the proteins - Bio source for nitrogen and sulfer -can be oxidized for energy in case of starvation	
Nucleotide s	- DNA and RNA - Trap and shuttle energy - ATP - signal transduction (turns hormone into a cellular response)	
Ochem v. Biochem		
Organic che	mistry is Bio chemistry is about	

carbon compounds that

crawl

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about carbon compounds