

### Lactate dehydrogenase

lactate + NAD<sup>+</sup> → pyruvate + NADH + H<sup>+</sup>  
tetrameric

H-form: aerobic, heart L→P

M-form: anaerobic, muscle/liver P→L

Reagents: Lactate, NAD<sup>+</sup>, Oxidized PMS,  
Oxidized NBT

Specific Activity stain

LDH-1: pyruvate inhibition

LDH-1/2: 2-hydroxybutyrate as S

LDH-4/5: greater heat stability

### Creatine Kinase

Creatine + ATP ↔ creatine phosphate +  
ADP + H<sup>+</sup>

Dimeric

Cardiac: MM+ MB (Myocardial infarction)

Skeletal: MM

Brain: BB

### Chymosin (Rennin)

Aspartic protease

Cleave single peptide bond,

release acidic C-terminal peptide

Ca induced aggregation of modified casein  
micelle → precipitate as curd

### Affinity label

Specific & Irreversible inhibitor

Specificity group & reactive group  
resembles substrate

TPCK on His-57 of Chymotrypsin

### Determination of enzyme activity

NAD<sup>+</sup>: absorbance change at 340nm

FAD: absorbance change at 440 nm X

### Deter

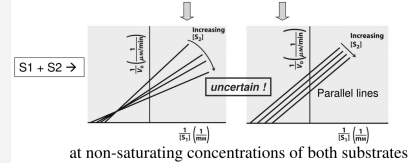
Follow INITIAL rate, rate drops

1. substrate depletion
2. reverse reaction
3. product inhibition
4. enzyme stability

### Differentiation

#### Differentiation of Mechanisms

- primary plot (single vs. double displacement)



- isotope exchange (single vs. double)
- product inhibition studies (specific type)

### Homeostatis(regulation)

1. [S] control, M-M vs Cooperativity
2. Allosteric effect
3. [S]cycle, 2-way, 6-Phosphofructokinase & Fructose biphosphatase
4. Zymogen activation
5. Covalent modification (phosphorylation, adenylation, myristoylation, ADP-ribosylation, methylation, acetylation, ubiquitination)
6. Enzyme cascade
7. Cascade amp
8. Enzyme induction/degradation

### Modification of amino acids

Active site residues more susceptible  
Ser-195 on chymotrypsin by DIPF → Act-ivity

Modify 1<sup>st</sup> with [S]/[I] to protect active site,  
then modify again in absence  
ADH inactivated by iodoacetate more than  
iodoacetamide

2AA involved: pKa >2 units apart: Good  
Close → tVmax never achieved

### Lite Beer

Barley: α-amylase cannot break down α-1,6  
bond + dextrin → Yeast

Glucoamylase from *Aspergillus niger*: break  
α-1,6 bond, less dextrin

### Aspartame

Thermolysin

**L-phenylalanine** + N-protected L-aspartate

N: benzyloxycarbonyl

### Catalysis

1. Strain/Distortion: entropy reduction
2. Acid-Base: carbonic anhydrase
3. Covalent catalysis: serine protease
4. Lower Ea: Zn&Arg127 in carboxypeptidase A stabilize TS

Histidine can be both e<sup>-</sup> donor/acceptor  
TS analogue: **pyrrole-2-carboxylate** on  
*proline racemase* as inhibitor  
Abzyme: mimic Ferrochelatase

C

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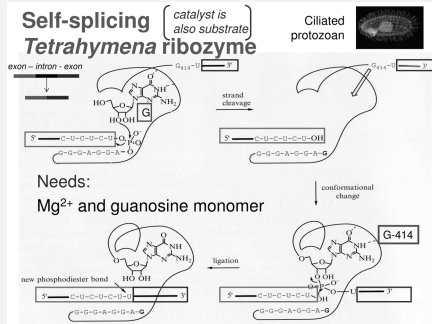
Page 1 of 3.

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## Ribozyme



RNA less versatile (4 building blocks AUCG)  
unable to form large non-polar region  
nucleic acid preferred as substrate  
RNA susceptible to hydrolysis

## Organic Cofactors

### Organic Cofactors

Coenzyme	Examples of chemical groups transferred	Dietary precursor in mammals
Biotin	CO <sub>2</sub>	Biotin Vitamin B <sub>7</sub>
Coenzyme A	Acyl groups	Vit B <sub>5</sub> Pantoic acid and other compounds
5'-Deoxyadenosylcobalamin (coenzyme B <sub>12</sub> )	H atoms and alkyl groups	Vitamin B <sub>12</sub>
Flavin adenine dinucleotide	Electrons	Riboflavin (vitamin B <sub>2</sub> )
Lipoate	Electrons and acyl groups	Not required in diet
Nicotinamide adenine dinucleotide	Hydride ion (H <sup>-</sup> )	Nicotinic acid (niacin) Vit B <sub>3</sub>
Pyridoxal phosphate	Amino groups	Pyridoxine (vitamin B <sub>6</sub> )
Tetrahydrofolate	One-carbon groups	Folate Vit B <sub>9</sub>
Thiamine pyrophosphate	Aldehydes	Thiamine (vitamin B <sub>1</sub> )

Catalytic cofactor: e.g. TPP/FAD  
Stoichiometric cofactor: cosubstrate

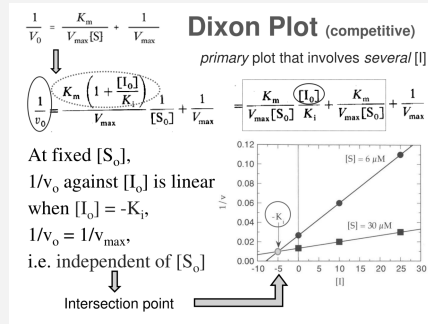
## Radioisotope

Glutamate decarboxylase  
CO<sub>2</sub> : trap gas in alkali  
Monoamine oxidase  
R-CHO: extracted by ether after acidification (acidified R-NH<sub>2</sub> will remain in aq phase)  
Cholinesterase  
COOH: ion exchange, importance of **label position**

## Scintillation Proximity Assay

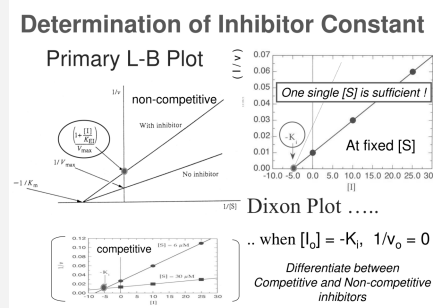
Radioligand stimulate bead to emit light, when in **close proximity**  
High affinity capture system: biotinylated substrate & streptavidin-coated beads  
NO separation needed. S or P bind to bead

## Competitive inhibitor



Same site, mutually exclusive  
Vmax unchanged Km increased

## Non-competitive inhibitor



ESI present, Km unchanged, Vmax decrease,  
equal Ki, same % inhib.

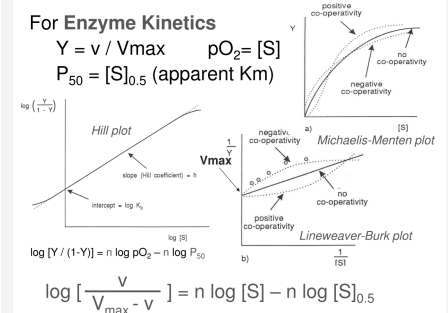
## Pre-steady state kinetics

E+S → ES (fluorescence)  
Stopped flow technique, follow time course of fluorescence change

## Irreversible inhibitor

Diisopropyl phosphofluoridate (DIPF)  
modifies *serine* on AchE

## Hill Coefficient



Important: Choice of [S]

Cooperativity: Same **site & ligand**

## Chymotrypsin

Active site **Ser195, His57 & Asp102** form charge relay system → high reactivity of Ser195  
Selective for carboxyl side of aromatic or large hydrophobic residue (Met)  
Biphasic kinetics:  
1. Burst phase: covalent complex  
2. SS Phase: hydrolysis + recovery  
Double displacement, p-nitrophenolate

Cofactor		
Apoenzyme	inactive form	
Cofactor	non-protein	
Holoenzyme		
Holoenzyme	active	
Metalloenzyme	participate in reaction (Lewis acid)	Zn in carbonic anhydrase
Metalloenzyme	stabilize transition state	Zn in carboxypeptidase A
Metal activated enzyme	maintain active conformation	K <sup>+</sup> in pyruvate kinase
Metal activated enzyme	form substrate complex bridge	Mg in kinase
Prosthetic group	tightly bound	NH <sub>2</sub> on pyridoxal phosphate of aspartate transaminase
Coenzyme		
Coenzyme	loosely bound	
Cosubstrate (coenzyme)	convert to product after Rx	NAD <sup>+</sup>

Coenzyme analogue as drug			
Drug	Analogue of	enzyme inhibited	MOA & Use
Sulfonamide	PABA	dihydropteroate synthase	folic acid synthesis, Abx
Methotrexate	Folate	dihydrofolate reductase	THF synthesis, childhood leukemia

### Plot

**Determination of Parameters**

$$v = \frac{V_{max} [S]}{K_m + [S]}$$

$$\frac{1}{V_0} = \frac{K_m}{V_{max} [S]} + \frac{1}{V_{max}}$$

$$V = -K_m \left( \frac{V}{[S]} \right) + V_{max}$$

**Lineweaver-Burk plot**  
double reciprocal

**Eadie-Hoast plot**

### Uncompetitive inhib.

S binding to E → expose site for I binding  
both Km Vmax decrease to same extent, ESI present, same slope

### Mixed inhibition

Binding affinity (Ki) not the same  
Vmax decrease, Km can increase/decrease

### Suicide substrate

P irreversibly bind to E  
**Deprenyl** on MAO on Flavin prosthetic group

### Substrate inhibition

High [S] favour ESS (nonproductive binding)  
e.g. succinate dehydrogenase (select points for drawing)

### Single displacement Rx

Random sequential creatine kinase  
Compulsory order ADH (NAD<sup>+</sup> bind 1<sup>st</sup>)  
**Ternary complex** present

### Double displacement Rx (Ping-Pong)

aspartate transaminase  
aspartate + α-ketoglutarate → oxaloacetate + glutamate (NH<sub>2</sub> displaced)

### Isotope exchange

Occurs only in double displacement; exception: maltose phosphorylase  
isotope from 1<sup>st</sup> P back to 1<sup>st</sup> S in absence of 2<sup>nd</sup> S e.g. sucrose phosphorylase  
Glu-Fru + Fru\* ↔ Glu-Fru\* + Fru

### Diff. subunits of multimeric enzyme

Catalytic & Regulatory	Aspartate transcarbamoylase (ATCase)	ATP & CTP
2 <sup>nd</sup> unit modify specificity	Lactose synthase	α-lactalbumin
2 diff. cat. units	tryptophan synthase (α2β2)	Tunnel connect active sites

### Lysozyme

hydrolyze glycosidic bond bet C-1 of NAM and C-4 of NAG, **Non-identical site**  
Site of cleavage: bet D & E, distorted Ring D  
Glu-35 as acid, H<sup>+</sup> to O of glycosidic bond  
Carbonium cation stabilized by  
1. -ve charge on Asp-52  
2. half-chair formation of sugar D (strain)  
(resonance stabilize +charge on C-1 with O)