

Lactate dehydrogenase

lactate + NAD⁺ → pyruvate + NADH + H⁺
 tetrameric
 H-form: aerobic , heart L→P
 M-form: anaerobic , muscle/liver P→L
 Reagents: Lactate,NAD⁺,Oxidized PMS,
 Oxidized NBT
 Specific Activity stain
 LDH-1: pyruvate inhibition
 LDH-1/2: 2-hydroxybutyrate as S
 LDH-4/5: greater heat stability

Creatine Kinase

Creatine + ATP ↔ creatine phosphate +
 ADP + H⁺
 Dimeric
 Cardiac: MM+ MB (Myocardial infarction)
 Skeletal: MM
 Brain: BB

Chymosin (Rennin)

Aspartic protease
 Cleave single peptide bond,
 release acidic C-terminal peptide
 Ca induced aggregation of modified casein
 micelle→ precipitate as curd

Affinity label

Specific & Irreversible inhibitor
 Specificity group & reactive group
 resembles substrate
 TPCK on His-57 of Chymotrypsin

Determination of enzyme activity

NAD⁺ : absorbance change at 340nm
 FAD: absorbance change at 440 nm X

Deter

Follow INITIAL rate, rate drops
 1. substrate depletion
 2. reverse reaction
 3. product inhibition
 4. enzyme stability

Differentiation

Differentiation of Mechanisms

- primary plot (single vs. double displacement)

at non-saturating concentrations of both substrates

- isotope exchange (single vs. double)
- product inhibition studies (specific type)

Homeostatis(regulation)

- [S] control, M-M vs Cooperativity
- Allosteric effect
- [S]cycle, 2-way, 6-Phosphofructokinase & Fructose biphosphatase
- Zymogen activation
- Covalent modification (phosphorylation,a-denylation,myristoylation,ADP-ribosylation,methylation,acetylation,ubiquitination)
- Enzyme cascade
- Cascade amp
- Enzyme induction/degradation

Modification of amino acids

Active site residues more susceptible
 Ser-195 on chymotrypsin by DIPF →Act-ivity
 Modify 1st with [S]/[I] to protect active site,
 then modify again in absence
 ADH inactivated by iodoacetate more than
 iodoacetamide
 2AA involved: pKa >2 units apart:Good
 Close→ tVmax never achieved

Lite Beer

Barley:α-amylase cannot break down α-1,6
 bond + dextrin → Yeast
 Glucoamylase from Aspergillus niger: break
 α-1,6 bond, less dextrin

Aspartame

Thermolysin
L-phenylalanine + N-protected L-aspartate
 N: benzyloxycarbonyl

Catalysis

- Strain/Distortion: entropy reduction
- Acid-Base : carbonic anhydrase
- Covalent catalysis: serine protease
- Lower Ea: Zn&Arg127 in carboxypeptidase A stabilize TS

Histidine can be both e⁻ donor/acceptor
 TS analogue: **pyrrole-2-carboxylate** on
proline racemase as inhibitor
 Abzyme: mimic Ferrochelatase

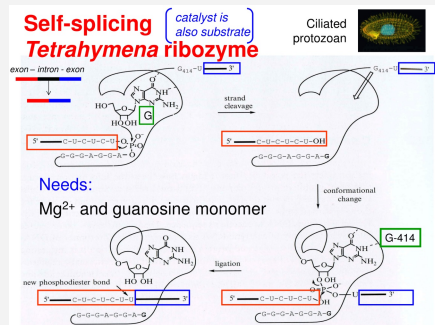


By **iplorip**
cheatography.com/iplorip/

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Ribozyme



RNA less versatile (4 building blocks AUCG)
unable to form large non-polar region
nucleic acid preferred as substrate
RNA susceptible to hydrolysis

Organic Cofactors

Organic Cofactors

Coenzyme	Examples of chemical groups transferred	Dietary precursor in mammals
Bioctylin	CO ₂	Biotin Vitamin B₇
Coenzyme A	Acyl groups	Vit B₅ Pantoic acid and other compounds
5'-Deoxyadenosylcobalamin (coenzyme B ₁₂)	H atoms and alkyl groups	Vitamin B₁₂
Flavin adenine dinucleotide	Electrons	Riboflavin (vitamin B ₂)
Lipoate	Electrons and acyl groups	Not required in diet
Nicotinamide adenine dinucleotide	Hydride ion (H ⁻)	Nicotinic acid (niacin) Vit B₃
Pyridoxal phosphate	Amino groups	Pyridoxine (vitamin B ₆)
Tetrahydrofolate	One-carbon groups	Folate Vit B₉
Thiamine pyrophosphate	Aldehydes	Thiamine (vitamin B ₁)

Catalytic cofactor: e.g. TPP/FAD
Stoichiometric cofactor: cosubstrate

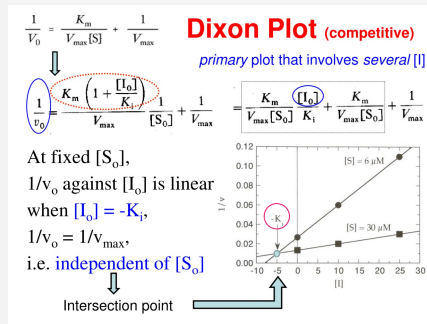
Radioisotope

Glutamate decarboxylase
CO₂: trap gas in alkali
Monoamine oxidase
R-CHO: extracted by ether after acidification (acidified R-NH₂ will remain in aq phase)
Cholinesterase
COOH: ion exchange, importance of **label position**

Scintillation Proximity Assay

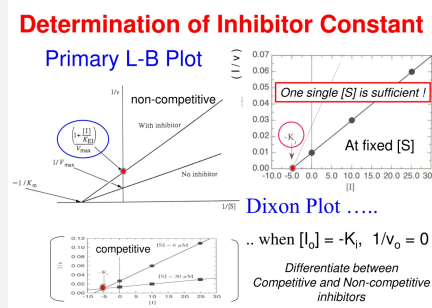
Radioligand stimulate bead to emit light, when in **close proximity**
High affinity capture system: biotinylated substrate & streptavidin-coated beads
NO separation needed. S or P bind to bead

Competitive inhibitor



Same site, mutually exclusive
Vmax unchanged Km increased

Non-competitive inhibitor



ESI present, Km unchanged, Vmax decrease,
equal Ki, same % inhib.

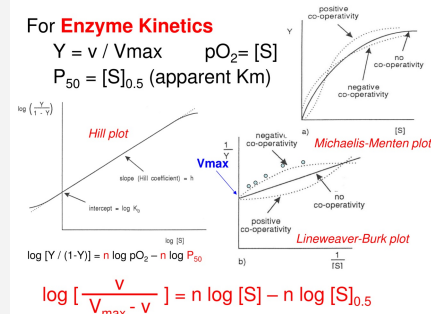
Pre-steady state kinetics

E+S → ES (fluorescence)
Stopped flow technique, follow time course of fluorescence change

Irreversible inhibitor

Diisopropyl phosphofluoridate (DIPF)
modifies *serine* on AchE

Hill Coefficient



Important: Choice of [S]

Cooperativity: Same **site & ligand**

Chymotrypsin

Active site **Ser195, His57 & Asp102** form charge relay system → high reactivity of Ser195

Selective for carboxyl side of aromatic or large hydrophobic residue (Met)
Biphasic kinetics:

1. Burst phase: covalent complex
 2. SS Phase: hydrolysis + recovery
- Double displacement, p-nitrophenolate

Cofactor		
Apoenzyme	inactive form	
Cofactor	non-protein	
Holoenzyme		
Holoenzyme	active	
Metalloenzyme	participate in reaction (Lewis acid)	Zn in carbonic anhydrase
Metalloenzyme	stabilize transition state	Zn in carboxypeptidase A
Metal activated enzyme	maintain active conformation	K ⁺ in pyruvate kinase
Metal activated enzyme	form substrate complex bridge	Mg in kinase
Prosthetic group	tightly bound	NH ₂ on pyridoxal phosphate of aspartate transaminase
Coenzyme	loosely bound	
Cosubstrate (coenzyme)	convert to product after Rx	NAD ⁺

Coenzyme analogue as drug			
Drug	Analogue of	enzyme inhibited	MOA & Use
Sulfonamide	PABA	dihydropteroate synthase	folic acid synthesis, Abx
Methotrexate	Folate	dihydrofolate reductase	THF synthesis, childhood leukemia

Plot

Determination of Parameters

$$v = \frac{V_{max} [S]}{K_m + [S]}$$

$$\frac{1}{V_0} = \frac{K_m}{V_{max} [S]} + \frac{1}{V_{max}}$$

$$V = -K_m \left(\frac{V}{[S]} \right) + V_{max}$$

Lineweaver-Burk plot
double reciprocal

Eadie-Hoast plot

Uncompetitive inhib.

S binding to E → expose site for I binding
both Km Vmax decrease to same extent, ESI present, same slope

Mixed inhibition

Binding affinity (Ki) not the same
Vmax decrease, Km can increase/decrease

Suicide substrate

P irreversibly bind to E
Deprenyl on MAO on Flavin prosthetic group

Substrate inhibition

High [S] favour ESS (nonproductive binding)
e.g. succinate dehydrogenase (select points for drawing)

Single displacement Rx

Random sequential creatine kinase
Compulsory order ADH (NAD⁺ bind 1st)
Ternary complex present

Double displacement Rx (Ping-Pong)

aspartate transaminase
aspartate + α-ketoglutarate → oxaloacetate + glutamate (NH₂ displaced)

Isotope exchange

Occurs only in double displacement; exception: maltose phosphorylase
isotope from 1st P back to 1st S in absence of 2nd S e.g. sucrose phosphorylase
Glu-Fru + Fru* ↔ Glu-Fru* + Fru

Diff. subunits of multimeric enzyme

Catalytic & Regulatory	Aspartate transcarbamoylase (ATCase)	ATP & CTP
2 nd unit modify specificity	Lactose synthase	α-lactalbumin
2 diff. cat. units	tryptophan synthase (α2β2)	Tunnel connect active sites

Lysozyme

hydrolyze glycosidic bond bet C-1 of NAM and C-4 of NAG, **Non-identical site**
Site of cleavage: bet D & E, distorted Ring D
Glu-35 as acid, H⁺ to O of glycosidic bond
Carbonium cation stabilized by
1. -ve charge on Asp-52
2. half-chair formation of sugar D (strain) (resonance stabilize +charge on C-1 with O)