

Biology 1 Cheat Sheet

by ggcheatgg (evelana) via cheatography.com/85968/cs/20068/

Water H2O - solvent of life

Water is the dominant component of virtually all living organisms, and most biochemical reactions

one oxygen and two hydrogen (connected via covalent bonds)

allows chemical reactions to occur inside living organisms

necessary for the formation of certain biological structures

polar molecule that can form hydrogen bonds - tetrahedral shape (because of four pairs of electrons on outer shell)

 $gas \rightarrow$ does not form hydrogen bonds

solid → molecules are held in a rigid state by hydrogen bonds

 $\label{eq:liquid} \begin{tabular}{ll} \textbf{liquid} \rightarrow \textbf{hydrogen bonds continually break} \\ \textbf{and form as water molecules move} \\ \end{tabular}$

Evapor-	Living systems use the evapor-
ation	ation of water, which disrupts
	hydrogen bonds, to dissipate
	excess heat that would
	otherwise cause problems.
Adhesion	the attraction of water molecules to other molecules of a different type
cohesion	the capacity of water molecules to resist coming apart from one

surface water molecules stick to one

tension

tension another

Macromolecules

Macromolecules: (=Polymers) Function: energy storage, structural support, transport, protection and defense, regulation of metabolic activities, means for movement / growth / development, heredity

Proteins The functions of proteins include support, protection (e.g., skin surface), catalysis, transport, defense, regulation, movement, signaling, and storage

Carboh- contain carbon bonded to
ydrates hydrogen and oxygen atoms and
have the general formula
(C1H2o1)n. act as energy
storage and transport molecules
and structural components

nucleic A polymer made up of nucleoacids tides, specialized for the
storage, transmission, and
expression of genetic information. DNA and RNA are nucleic
acids.

Nonpolar, hydrophobic molecules that include fats, oils, waxes, steroids, and the phospholipids that make up biological membranes



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another when places under

Not published yet. Last updated 18th July, 2019. Page 1 of 1.

lipids

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