

Naming	Formation Reactions	Other Reactions
Alkane -ane	Forming Alcohols Alkene + H <sub>2</sub> O = (H <sub>2</sub> SO <sub>4</sub> ) Alcohol	Combustion RX Complete : __ + O <sub>2</sub> = CO <sub>2</sub> + H <sub>2</sub> O
Alkene -ene	HYDROGENATION RX Aldehyde (reducing agent) = 1 alcohol	Incomp: __ + O <sub>2</sub> = CO <sub>2</sub> + H <sub>2</sub> O + CO + C
Alkyne -yne	Keytone (reducing agent) = 2 alcohol	Alkanes SUBSTITUTION RX
Cyclic cyclo-	Forming Ethers CONDENSATION RX	Alkenes & Alkynes ADDITION RX
Aromatics -benzene	Alcohol + Alcohol (H <sub>2</sub> SO <sub>4</sub> ) = Ether + H <sub>2</sub> O	Halogenation... Hydrogenation... Hydrohalogenation... Hydration
Alcohol -ol	Forming Aldehydes OXIDATION RX	Esters HYDROLYSIS (soaponification)
Ether -oxy- or -ether	1 Alcohol + [O] = Aldehyde + H <sub>2</sub> O	Ester + NaOH = Sodium Carboxylate + Alcohol
Aldehyde -al	Forming Keytones 2 Alcohol + [O] = Keytone + H <sub>2</sub> O	Polarity
Keytone -one	Forming Carboxylic Acids Aldehyde + [O] = Carboxylic Acid	Carboxylic Acid
Carboxylic Acid -oic acid	Forming Esters (esterification) Carboxylic Acid + Alcohol (H <sub>2</sub> SO <sub>4</sub> ) = Ester + H <sub>2</sub> O	Alcohol
Ester -oate	Forming Amines Alkyl Halide + Ammonia = Amine + HI	Amines & Amides
Amine -amine or amino-	Forming Amides Carboxylic Acid + Ammonia (H <sub>2</sub> SO <sub>4</sub> ) = Amide + H <sub>2</sub> O	Aldehydes, Keytones, Esters
Amide -amide		Ethers & Alkyl Halides
Thiol -thiol		Alkenes & Aromatics
Formation Reactions		Alkanes
Forming Alkanes		
Hydrogenation of <i>Alkene</i> (Addition RX)		
H <sub>2</sub> + Alkene = Alkane		
Forming Alkenes		
Dehydration of <i>Alcohol</i>		
Alcohol (H <sub>2</sub> SO <sub>4</sub> ) = Alkene + H <sub>2</sub> O		
Substitution RX + Elimination RX		
Alkane + Halogen = Alkyl Halide + (HCl)		
Alkyl Halide + OH <sup>-</sup> = Alkene + (Cl) + H <sub>2</sub> O		

By emilyaltmann

[cheatography.com/emilyaltmann/](https://cheatography.com/emilyaltmann/)

Published 19th September, 2019.

Last updated 19th September, 2019.

Page 1 of 1.

Sponsored by [CrosswordCheats.com](http://crosswordcheats.com)

Learn to solve cryptic crosswords!

<http://crosswordcheats.com>