

Energy (Thermal Energy) Cheat Sheet

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Energy (Temperature Changes)

Internal (Thermal) Energy -> Energy stored in the movement of particles Amount of internal energy depends on temperature, materials, mass Temperature -> How hot or cold something is

Efficiency Formula (Energy)

Formula: Efficient Powers / Total 4/40 x

Amount of Electricity Produced x 100%

100% = Percentage = 100

Energy Formula

Formula: Q = Q -> Amount of heat mcAT transfered

Energy Formula

Formula: Q = mcAT Q -> Amount of heat transfered

M = Mass C = Specific Heat Capacity

A = The changes in T = Temperature temperature

In a refrigerator, 2kg of water cools from 30 °C to 0 °C and then freezes to form ice at 0 °C. The specific heat capacity of water is 4200 J kg °C, and the specific latent heat of fusion of ice is 336000 J kg °C. What is the heat washed in this process? $Q1 = mcAT = 2 \times 4 \times 200 \times -30$ $Q2 = mAHf = 2 \times -336000$

Listric Energy

Q1 + Q2 = -924000

Appliances -> Machines	Power -> Different appliance transfer different amount per second
Wratts -> Method measurement of the rate of electricity	Power rattings -> Max amount of power a device produce under normal circumtances
Efficiency -> Ratio of useful energy from total energy	Sankey Diagram -> Diagram that shows efficiency

Transfering Energy

Paying for Energy

Unit (Times, Electricity and gas are Energy, Power paid base on usage and wrattings) measure in KwH

Formula -> KW x H x KwH (Hour)

Controlling Energy Transfer

Insulation -> Retain warmth Air is only inside the house and good reduces fuel costs. conductor if it is trapped To test insulation effective-Light color -> Reflect ness, use containers with different insulating materials energy, and measure temperature while Dark changes over time. color -> Emit radiation

Evaporation	Conduction -> Passing vibrations (heat)
Radiation -> All objects emit infrared radiation (hotter, emit more, no need medium, thermal imagers)	Convection - > Cooler fluid sink, hotter fluids
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