

### Chapter 1

#### Scientific Method

- Make observations
- Make a hypothesis
- Has to be tested
- Make prediction assuming hypothesis is true
- Carry out experiments
- Mathematics is the tool to understand and predict the natural world

#### Scientific Method and Meteorology

- Atmosphere obeys laws of physics and chemistry
- Instruments allow us to quantify the state of the atmosphere
- Thermometer
- Hygrometer
- How much humidity/moisture in air
- Barometer
- Air pressure
- Anemometer
- Wind speed/wind direction
- Mathematics can project current conditions into the future
- Uses computer models to help with calculating

#### Weather and Climate

- Weather describes state of the atmosphere at any given time
- Temperature
- Air Pressure
- Humidity
- Cloud Cover
- Precipitation
- Visibility
- Wind Velocity

### Chapter 1 (cont)

- Climate describes average atmospheric conditions over at least 30 years
- Includes extremes
- The frequency of extremes help differentiate between locations with similar averages

#### Meteorology

- Meteorologica
- Book on natural philosophy by Aristotle in 340 BC
- How meteorology got its name
- Study of the atmosphere and its phenomena
- Began with the invention of weather instruments (1450--1650)
- Quantified the atmosphere
- Allows for the prediction of what the atmosphere will do
- Telegraph (1843)
- Allowed for the transmission of current weather conditions across vast areas
- Weather Map Analyses (1869)
- Visual snapshot of current state of the atmosphere
- Understanding of Air Masses and fronts (1920)
- Key weather features that drive world weather patterns
- Daily weather balloon launches (1940s)

### Chapter 1 (cont)

- These radiosondes provide 3D view of the atmosphere
- Numerical Weather Prediction (NWP)
- Solving the mathematical laws of physics/chemistry at high speeds

#### Remote Sensing of the Atmosphere

- Weather Radar (1940s)
- Detects precipitation targets from over 100 miles away
- Doppler Weather Radar (1990s)
- Detects precipitation targets and their motion
- "Sees" the wind
- Dual Pole Doppler Weather Radar (2000s)
- Distinguishes between rain, snow, hail, and bugs
- Weather Satellites (1960s)
- Reveal weather features produced by cloud patterns
- Can supply NWP with data from every location on Earth
- Most Common Type of Satellite
- Geostationary
- Orbits the Earth at the same speed the Earth spins
- GOES 16 and 17 Satellites
- Best View of the US

### Chapter 1 (cont)

- Centered over the Equator (0° Latitude) and 75° W | 137°W Longitude
- 22,300 Miles

#### Latitude

- the angle made between center of Earth and a point on surface using the Equator as the reference line
- North Pole = 90° N Latitude | South Pole = 90°S Latitude
- Most of the US is between 30°N and 50°N Latitude

#### Longitude

- the angle made between center of Earth and a point on surface using the Prime Meridian as the reference line
- Runs from N Pole to S Pole through Greenwich, England
- Most of the US lies between 70°W and 125°W Longitude

#### Most Common Type of Storm System

- Middle-Latitude Cyclonic Storm System
- Extratropical Cyclone
- Cyclone=area of low pressure
- Anticyclones=area of high pressure

#### Depiction of Winds

- Wind is defined from the direction it is blowing



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Page 1 of 11.

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### Chapter 1 (cont)

- Represented by a line - *wind barb* - drawn parallel to wind
- Points in direction from which the wind is blowing
- Speed is represented by wind flags
- Full flag = 10 knots
- Half flag = 5 knots

#### Wind Flow

- Counterclockwise and inward around lows
- Clockwise and outward around highs

In Southern hemisphere...

- Clockwise/inward around lows
- Counterclockwise/outward around highs
- Wind does not cycle around the equator
- No hurricanes but tornados can still happen

#### Vertical Wind Flow Around High and Lows

- Air converges and rises in the center of low pressure (cyclone)
- Clouds/precipitation
- Air diverges and sinks in the center of high pressure (anticyclone)
- Clear skies

#### Weather Fronts

- Cold front
- Boundary that separates colder air from warmer air
- When colder air advances and replaces warmer air

### Chapter 1 (cont)

- Warm front
- Boundary that separates colder air from warmer air
- When warmer air advances and replaces colder air
- Occluded front
- When cold front merges with warm front

#### Impacts of Weather and Climate

- Weather dictates the clothes we wear on any given day

#### Impacts of Weather and Climate

- Climate dictates the clothes we have in our wardrobe
- Climate dictates the type of crops we can grow
- Weather dictates whether the crops can be harvested

#### When Weather is Not What it Seems

- Wind Chill
- Body perceives a lower temperature than it really is
- Hypothermia
- Frostbite
- Heat Index
- Body perceives a higher temperature than it really is
- Hyperthermia
- Heat Exhaustion or Heatstroke

#### Other Biological Impacts

### Chapter 1 (cont)

- Rapid pressure falls/rising humidity
- Can induce expansion of joints and cause joint pain
- Wind flowing downhill heats up
- Chinook winds/ Santa Ana
- Incidence of depression increases

#### Economical Impacts of Weather

- Warm winters = lower heating bills
- Beware of unusual winter severe weather outbreaks
- Cold winters = higher heating bills
- Severe Artic air cold snaps can threaten human lives/infrastructure damage/massive crop losses
- Heat Waves and Drought
- Crops losses
- Wildfires increase
- #1 in weather-related fatalities

#### Climate Change Bringing More Extremes

- Heat waves and drought increasing
- Flooding events increasing
- Hurricane intensity increasing

#### Other Weather Hazards

- Severe Thunderstorms
- 50 knot (58 mph) winds
- 1-in hail
- Tornado

### Chapter 1 (cont)

- It has to fulfill one condition to be considered
- Flash Flooding
- Slow-moving thunderstorms
- "Training" storms
- Downburst winds
- Macrobust- greater than 4 km (2.5 mi) in diameter
- Microburst - less than 4 km in diameter
- Both produce Wind Shear
- Change in wind speed/direction over short distance

#### Who Studies This?

- Meteorologist
- Professionally trained, college degree in atmospheric science
- Weathercaster
- Good communicator of weather information

#### Weather Business is Expanding

- Private Meteorological
- App Development
- Forensic services

#### Fundamentals of Meteorology

- The atmosphere is a mixture of gases
- Nitrogen
- Oxygen
- Argon
- Water Vapor (highly variable)
- Carbon Dioxide (generally increasing)



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Page 2 of 11.

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### Chapter 1 (cont)

Most of the gases are near the surface

- Air gets "thinner" as you go up
- 99% of atmosphere is within 19 mi of the surface
- Blocks deadly solar radiation from reaching the surface

**The First Atmosphere**

- 4.6 BYA the atmosphere was mostly hydrogen and helium
- Some methane and ammonia thrown in
- Hydrogen and helium escaped into space
- Earth's gravity not strong enough

**The Second Atmosphere**

- Outgassing from Earth's hot interior (volcanoes)
- Mostly water vapor (80%), carbon dioxide (10%), and some nitrogen
- Water vapor "condensed" into clouds with rain lasting 1000s of years
- Combined with asteroid/comet collisions that formed the oceans

*Water Vapor Levels in Atmosphere Drop*

- Most of the water vapor converted to liquid water
- Atmosphere now just a few % water vapor

*CO2 Levels Drop*

- CO2 readily dissolves in water

### Chapter 1 (cont)

- Combined with chemicals in the ocean to form limestone

*N2 Levels Increase*

- Nitrogen is not very chemically reactive
- Once in the atmosphere, tends to stay

*O2*

- Solar radiation splits water vapor into hydrogen and oxygen
- Hydrogen escaped into space
- Oxygen left behind
- 2.4 BYA, something wonderful happens
- Cyanobacteria (blue-green algae) produce O2 from photosynthesis

- Earth begins to cool
- O2 combines with O to form O3 (ozone)
- O3 in upper atmosphere absorbs incoming radiation cooling the Earth
- Methane breaks down in the presence of O2
- Warming effect of methane weakens
- Earth becomes very cold
- Cyanobacteria proliferate around the world removing CO2 from atmosphere
- Warming effect of carbon dioxide weakens

**First Mass Extinction**

- Earth gets covered in ice
- Frigid Earth no longer supporting life
- No O2 being produced

### Chapter 1 (cont)

- O2 is highly reactive/combines with other elements to form rocks
- O2 levels drop worldwide

**Life Gets a Second Chance**

- Outgassing increases water vapor/carbon dioxide (volcanoes)
- H2O and CO2 are warming gases
- Takes over a billion years for new photosynthesizing life to reappear
- After another 1/2 billion years, O2 levels are where they are today

**Water Vapor**

- 0% to 4% of the atmosphere
- Always invisible
- Become visible when vapor molecules "jump" on each other to form droplets or ice crystals
- Condensation form droplets
- Deposition form ice crystals
- Evaporation is when liquid turns to gas
- Water is only substance that can exist in all 3 phases at normal temperature/pressure

**Characteristics**

- Greenhouse Gas
- Very effective at absorbing outgoing radiation emitted by Earth

### Chapter 1 (cont)

- Re-emits some of this energy back keeping the Earth warmer

**Carbon Dioxide**

- Greenhouse Gas
- Comes from the decay of vegetation
- Volcanic eruptions
- Burning of coal, oil, natural gas (fossil fuels)
- Removed by photosynthesis of land and ocean plants
- CO2 gets stored in roots, branches, and leaves
- Chemical weathering of rocks
- CO2 dissolves in rainwater
- Forms carbonic acid
- Combines with minerals in rocks and becomes part of the rock
- Dissolves in Ocean water
- Used by sea critters to make shells
- Eventually sinks to bottom of the sea

**Vertical Structure of the Atmosphere**

Air is compressible

- Gravity pulls most -but not all- air molecules near the surface
- Air Density = # of air molecules in a given volume
- Mass/volume

Air Density and Air Pressure



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Page 3 of 11.

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### Chapter 1 (cont)

- Force exerted by air molecules against a surface
- Same thing as the weight of the air above you
- At sea level, air weighs 14.7 lbs per square inch

#### Measuring Air Pressure

- 14.7 lbs/in<sup>2</sup>
- 1013.25 millibars (mb)
- 29.92 in Hg

#### Atmosphere is Very Thin

- Half of the air molecules is below 5.5 km (18,000 ft)
- 99.9% is below 50 km (160,000 ft)

#### Layers of the Atmosphere

- Atmospheric layers are defined by how the temperature changes with height
- Lapse Rate= rate at which temperature decreases with height
- In lower atmosphere, lapse rate = 6.5°C per km (3.6°F per 1000 ft)
- Temperature can INCREASE with height
- This is called temperature inversion
- Lapse rate is negative
- Troposphere
- Stratosphere
- Mesosphere
- Thermosphere
- Ionosphere
- Lower part of the Thermosphere

### Chapter 1 (cont)

- Solar energy strips electrons from N<sub>2</sub> and O<sub>2</sub> causing them to glow

#### Northern and Southern Lights

- Aurora Borealis - N. Lights
- Aurora Australis - S. Lights

### Chapter 2

- Energy
- The ability to do "work": when an object moves
- Kinetic Energy
- Energy of motion- translational, rotational and vibrational
- Potential Energy
- Energy that can convert to kinetic energy
- Water that is behind a dam
- Object suspended in the sky
- Temperature
- Average kinetic energy of atoms in a substance
- Some move fast, others not so fast
- Average motion = temperature
- When molecules move, rotate, and/or vibrate, we say that the object has a temperature
- When air molecules move slowly, they crowd together
- It is cold and air is dense

### Chapter 2 (cont)

- When air molecules move quickly, they spread out
- We say it is warm and the air is less dense
- Internal Energy (Heat energy)
- The total kinetic and potential energy of all atoms or molecules
- Heat
- Heat is the transfer of energy from warmer objects to cooler ones
- The bigger the temperature difference, the faster the energy transfer
- Know how to convert temperature scales
- ## Temperature Measurements
- Important Temperature Values
- Ice point
- Ice melts, water freezes
- 32°F, 0°C, 273.15K
- Steam point
- Water boils
- 212°F, 100°C, 373.15 K
- ## Types of Heat Energy
- Sensible Heat
- Heat energy that can be absorbed or released by a substance that results in a change of temperature
- Latent Heat

### Chapter 2 (cont)

- Heat energy that is absorbed or released by a substance when the substance undergoes a phase change
- Temperature of substance does not change
- Let ice at 32°F absorb heat energy
- Ice melts, but its temp remains at 32°F
- Only after the ice completely melts will the water warm up
- If water freezes, it releases the same heat it took to cause it to melt in the first place but water temp does not change
- The surrounding air does warm
- How does heat energy get transferred?
- Conduction
- Heat transfer by contact of one substance with another
- Energy gets transferred from one molecule to the next
- Some materials transfer heat better than others
- Metals are good conductors
- Fiberglass, cork, wood, cloth, glass, water are poor conductors
- Air is a poor conductor of heat
- Heat (Thermal) Conductivity



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Page 4 of 11.

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### Chapter 2 (cont)

- Measure of how well a substance transfers heat energy
- ## Why Temperature Decreases with Height
  - Atmosphere is mostly "transparent" to incoming sunlight
  - Sun does not heat air directly
  - Atmosphere can be heated by conduction
  - Ground absorbs sunlight
  - Air in contact with ground gets heated
  - Heat energy does not get conducted to higher altitudes very well
- ## How Does Heat Energy Get Transferred?
  - Convection
    - Heat transferred due to the movement of a substance from one place to another
    - Much more efficient than conduction
    - Moving heat energy vertically and horizontally
    - Convection - vertical air motions, also called thermals
    - Advection - horizontal air motions
  - Radiation (Electromagnetic Radiation or Radiant Energy)
    - The only heat transfer possible in a vacuum
    - Can also transfer heat in air or water

### Chapter 2 (cont)

- Energy is carried by photon particles defined by their wavelength
- nm=nanometer= billionth of a meter
- $\mu\text{m}$ =micrometer or micron= millionth of a meter
- ## Radiation Laws
  - Facts
    - All objects above Absolute Zero radiate (emit) energy at ALL wavelengths
    - $0\text{K} = -273.15^\circ\text{C} = -459.67^\circ\text{F}$
    - Even in interstellar space, the temperature is between 2.7K and 5K
    - Total Radiation emitted = Sum of energy emitted from every wavelength
    - Total Energy emitted by every square meter of an object is given by the Stefan-Boltzmann Law
      - $E = \sigma \times T^4$  (results is in Watts per square meter,  $\text{W/m}^2$  or  $\text{W m}^{-2}$ )
    - Watts=Joules per second, J/s or  $\text{Js}^{-1}$
    - Temperature has to be in Kelvins
    - $\sigma$  is constant variable
    - Hot objects emit more radiation than cooler objects
    - There is one wavelength ( $\lambda_{\text{max}}$ ) that an object will emit most of its radiation
    - Wien's Displacement Law

### Chapter 2 (cont)

- $\lambda_{\text{max}} = 2897/T$  (answer is in  $\mu\text{m}$ )
- T must be expressed in Kelvin
- Stefan-Boltzmann and Wien's Law
  - Only valid if object is a blackbody object
  - An object that is a perfect absorber and perfect emitter of radiation
  - Absorbs all radiation that strikes it and then emits max possible radiation
  - Absorption and Emission
    - If an object absorbs radiation, it must also emit radiation
    - If absorption is greater than emission, object heats up
    - If absorption is less than emission, object cools down
    - If absorption = emission, object's temperature remains the same
  - Radiative Equilibrium Temperature
  - Absorption
    - Gas molecules are picky about which type of radiation they will absorb
    - Selective Absorbers
      - Some will only eat shortwave radiation
      - Shortwave is less than  $1.4\mu\text{m}$
      - Some will only eat longwave radiation

### Chapter 2 (cont)

- Longwave is greater than  $1.4\mu\text{m}$
- Some will eat both or neither
- Air is mostly transparent to incoming solar radiation
- Air is not transparent to outgoing terrestrial radiation
- Some gases absorb Earth's outgoing radiation and then reemit some of it back to the surface
- ## Main Greenhouse Gases
  - Water Vapor
  - Carbon Dioxide
  - Methane
  - Nitrous Oxide
  - Ozone
- ### Benefit of the Greenhouse Effect
  - Average temperature of the Earth is  $59^\circ\text{F}$
  - Without greenhouse gases, it would be  $0^\circ\text{F}$
- ## What Else Happens to Radiation When it Enters the Atmosphere?
  - Transmission
    - Radiation passing through air molecules without interacting with any of them
    - About 55% of incoming solar radiation is transmitted
  - Reflection
    - Radiation that bounces off an object at the same angle object, and it leaves at the same intensity



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Page 5 of 11.

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### Chapter 2 (cont)

- Scattering
- Produces a large number of rays traveling in all different directions
- Scattered radiation is weaker than what originally hit the object
- Gasses scatter solar radiation preferentially
- Some wavelengths are scattered better
- Atmospheric gases - mostly N<sub>2</sub> and O<sub>2</sub> scatter blue/violet more effectively than reds/oranges
- Violet is scattered best
- Blue is at higher intensity level than violet
- Human eye detects blue better than violet
- When Sun is on horizon, light travels through a lot more atmosphere than when Sun is overhead
- ## Quantifying Reflected Radiation
- Albedo
- Percentage of radiation reflected by an object
- Average albedo for Earth is about 30 percent
- Average albedo for the Moon is about 7 to 12 percent
- A perfect reflector would have an albedo of 100%
- Saturn's moon Enceladus has an albedo of almost 100%
- ## What Causes Temperature Differences?

### Chapter 2 (cont)

- Solar Radiation Intensity largely determines temperature
- High solar radiation intensity = tropical areas
- Low solar radiation intensity = arctic/antarctic area
- Solar Radiation Intensity = Power/Area
- Partly determined by height of Sun above horizon
- Midday sun always high in the sky in the Tropics
- Midday sun never high in the sky in Arctic/Antarctic areas
- ## Important Dates to Remember
- Solstices and Equinoxes
- Summer solstice: June 21 or 22
- At solar noon, sun's rays are vertical at Tropic of Cancer 23 1/2° N Latitude
- Longest day
- Winter solstice: December 21 or 22
- Vertical rays at Tropic of Capricorn: 23 1/2° S Latitude
- Shortest day
- Autumnal equinox: September 22 or 23
- Vertical rays at the equator
- day and night are equal

### Chapter 2 (cont)

- Vernal (spring) equinox: March 21 or 22
- Vertical rays at the equator
- 12 hour days/nights everywhere
- ## Solar Noon
- The time when the sun reaches its highest point in the sky
- halfway between sunrise and sunset
- ## Solar Declination
- The latitude where Sun is directly overhead at solar noon
- Can only be in the tropics on any given day
- ## Solar Elevation Angle (SEA)
- Angle the Sun makes with horizon at any time
- When sun is on horizon, SEA is 0°
- Halfway up into the sky, the SEA is 45°
- If sun is directly overhead, SEA is 90°
- ## Solar Noon Angle (SNA)
- Angle the Sun makes with the horizon at solar noon
- Sun will be at its highest point in the sky at solar noon on any given day
- Sun will be at absolute highest point in the sky at solar noon on the first day of summer

### Chapter 2 (cont)

- Sun will be at its absolute lowest point in the sky at solar noon on the first day of winter
- ## Procedure to Finding the Solar Noon Angle

  1. Where is the Solar Declination
  2. Calculate latitude difference to the SD
  3. Subtract this difference between 90°

### Chapter 3

- ## Temperature Definitions
- Daily Mean
- Average of the 24 hourly-temperature readings
- Add high | low and divide by two
- Daily Temperature Range
- Difference between high and low
- Monthly Mean
- Average of daily means for the month
- Annual Mean
- Average of the 12 monthly-means for the year
- Annual Temperature Range
- Difference between highest and lowest monthly mean
- ## Controls of Temperature
- Anything that determines the temperature of a location is called "Control of Temperature"
- #1 Control: Amount of Solar Radiation received



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Page 6 of 11.

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### Chapter 3 (cont)

- Solar Angle
- Length of daylight
- Latitude determines solar angle and day length
- Areas on same latitude have the same solar angles and number of daylight hours on any given day
- So latitude is no. 1 control
- Differential heating of land and water
- Land and water do not heat up/cool down at the same rate
- Water requires 3-5x as much energy than land to heat up to the same temperature
- Geographic position
- If prevailing winds blow from sea to land (windward coast), temperatures will not change much
- Tend to have a small annual temperature range
- If prevailing winds blow from land to sea (leeward coast), temperatures will fluctuate much more
- Tend to have a larger annual temperature range
- Ocean currents
- Warm Gulf Stream
- "River" of warm water transports heat to northern latitudes
- Western Europe is much milder than it should be
- Effects of warm ocean currents
- Palm Trees in England and Ireland

### Chapter 3 (cont)

- Effects of Cold/Warm Ocean Currents
- Poleward currents bring warmer conditions
- Equatorward current bring cooler conditions
- Another Effect of Cold Ocean Currents
- Land is pretty dry
- Ex. Atacama Desert is the driest place on Earth due to the cold Peru Ocean Current
- Elevation
- Temperatures usually decrease with altitude in the troposphere
- Atmosphere is mostly transparent to solar radiation but the ground absorbs almost all radiation that hits it
- Air in contact with ground (conduction) heats up the most
- Atmosphere is heated from bottom up
- Cloud cover
- Clouds (or water vapor) lower surface temperatures during the day
- Clouds (or water vapor) increase surface temperatures at night
- Albedo variations
- High albedo reduces surface temperature
- Low albedo increases surface temperature
- ## Consequences of Being Next to Large Body of Water

### Chapter 3 (cont)

- Inland Winnipeg and Coastal Vancouver
- Cities are located at a similar latitude
- Vancouver has a milder climate
- Temperatures in S. hemisphere do not fluctuate as much
- S. Hemisphere is the "water hemisphere"
- Water moderates temperatures
- ## Minimum and Maximum Temperatures of the Day/Year
- Delay in reaching high temperature: Lag of the Maximum
- Also applies to the seasons
- ## Urban Heat Island Effect
- Interior sections of cities tend to be warmer than surrounding rural areas
- ## Temperature Measurements
- Mechanical thermometer
- Liquid in glass | expands when heated, contracts when cooled
- Maximum thermometer - mercury
- Minimum thermometer - alcohol
- Thermograph
- Two metals in the strip will expand/contract differently depending on the temperature

### Chapter 3 (cont)

- Electrical thermometers
- Thermistor measures the resistance to electric current
- Provides accurate temperature reading even when temperature changes quickly (radiosondes)
- Instrument shelters
- White box
- Louvered sides (slits in the housing unit)
- Over grass and away from buildings
- 1/5m (5 feet) above ground
- ## Crop Protection Against the Cold
- Frost/Freeze Prevention
- Water sprinklers add heat from the latent heat of fusion when the water freezes
- Air mixing uses wind machines to mix warm and cool air
- Orchard heaters produce the most successful results, but fuel cost and pollution can be significant
- ## Heat Stress and Wind Chill: Indices of Human Discomfort
- Heat Stress Index: Temperature the body perceives when you include effects of humidity
- Evaporation of sweat is reduced when humidity is high



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Page 7 of 11.

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### Chapter 3 (cont)

- Apparent Temperature: temperature a person perceives
- Wind Chill Index: Temperature the body perceives when you include effects of wind
- Cold, dry air will evaporate moisture from the body
- Wind will blow away isolating air layer that surrounds the body
- ## Temperature and the Economy
- Heating Degree Days
- Gives a sense of how often one needs to heat a building
- Assumption: Heating not required when daily mean temperature is  $\geq 65^{\circ}\text{F}$ , then there is zero heating degree days
- For each degree the mean temperature  $< 65^{\circ}\text{F}$ , this is counted as one Heating Degree Day
- Cooling Degree Days
- Cooling not required when daily mean temperature is  $65^{\circ}\text{F}$  or lower
- Each degree of temperature  $> 65^{\circ}\text{F}$  is counted as one Cooling Degree Day
- Heating/Cooling Degree Days and Electric Bills
- Heating bill correlates with heating degree days
- Growing Degree Days

### Chapter 3 (cont)

- A way to determine if a crop can be successfully grown in any given area
- Assumes a BASE temperature for any given crop
- If daily mean is below the base, the plant goes dormant
- The difference between this BASE temperature and the daily mean temperature is a Growing Degree Day
- Some crops go dormant if daily mean is too high
- If daily mean  $> 86^{\circ}\text{F}$ , many plants stress out, go dormant
- In this event, the number of GDD is set at zero

### Chapter 4

- ## Water: A Unique Substance
- ### Hydrogen Bonding
- The attractive force between  $\text{H}_2\text{O}$  molecules
- Hydrogen side of  $\text{H}_2\text{O}$  is "+" charged | Oxygen side is "-" charged
- + Hydrogen side attracted to oxygen side of other  $\text{H}_2\text{O}$  molecules
- ### Formation of Ice
- When it's cold,  $\text{H}_2\text{O}$  molecules cannot break their bonds
- Remain fixed in a crystalline structure, ice
- Lowest kinetic energy state
- ### Liquid Water

### Chapter 4 (cont)

- When it's warmer,  $\text{H}_2\text{O}$  molecules break bonds temporarily
- Flow over each other but remain connected, liquid water
- Higher kinetic energy state
- ### Water Vapor
- When it's very warm,  $\text{H}_2\text{O}$  molecules break bonds completely
- Molecules scatter in random directions, gaseous water = water vapor
- Highest kinetic energy state
- ### Ice-Water-Water Vapor
- When water absorbs or releases internal energy, it can change phase
- ## Heat Energy
- One calorie of heat energy is required to raise 1 gram of water  $1^{\circ}\text{C}$
- ## Water: Changing Phases
- Latent Heat of Melting: 80 calories
- 80 calories of heat absorbed by 1g of ice melts into 1g of water
- No temperature change in the ice, but surrounding air gets colder
- Latent Heat of Fusion: 80 calories
- 80 calories of heat released by 1g of liquid water freezes to 1g of ice

### Chapter 4 (cont)

- No temperature change in the water but surrounding air does heat up
- Latent Heat of Vaporization: between 540 and 600 calories
- 540 to 600 calories absorbed by 1g of liquid water evaporate to 1g water vapor
- Heat is taken from surrounding air resulting in decrease air temperature
- Latent Heat of Condensation: between 540 and 600 calories
- 540 to 600 calories released by 1g water vapor condenses to 1g liquid water
- Heat is added to the surrounding air resulting in a temperature increase
- Latent Heat of Sublimation: about 680 calories
- 680 calories of heat absorbed by 1g of ice to sublime to 1g of water vapor
- Heat removed from surrounding air greatly cools the air around the ice
- Latent Heat of Deposition: about 680 calories
- 680 calories of heat released by 1g of water vapor deposits to 1g of ice
- Heat added to the environment greatly warms the air around the ice



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Page 8 of 11.

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### Chapter 4 (cont)

#### ## Measuring Water Vapor

- Mixing Ratio
- Mass of water vapor/mass of dry air it is in
- Water vapor measured in grams
- Dry air measured in kg
- Actual mass of water vapor in the air
- Saturation Mixing Ratio
- Max amount of water vapor allowed in the air (mostly determined by temperature)
- Vapor Pressure
- Pressure exerted by water vapor
- Total air pressure = sum of pressure from each gas
- The more water vapor in the air, the greater its contribution to the total air pressure
- Relative Humidity
- Mass of water vapor/mass of water vapor allowed to be in the air
- $RH = \text{mixing ratio} / \text{saturation mixing ratio}$
- $RH = \text{vapor pressure} / \text{saturation vapor pressure}$
- Relative humidity can be misleading
- Does not tell you how much water vapor is in the air unless you know the temperature of the air
- RH does tell you how close you are to saturating the air

### Chapter 4 (cont)

- When air is saturated, condensation occurs
- The lower the RH, the faster water evaporates
- Watering lawn in the morning is more effective than watering in the afternoon
- RH can be over 100% but not for long (supersaturation)
- Violent updrafts in thunderstorms can supersaturate
- Dew Point Temperature (Td)
- Temperature at which saturation occurs
- Better way of measuring actual water vapor content
- Absolute Humidity
- Specific Humidity
- #### Relationship Between T, Td, and RH
  - When T and Td are close, RH is high
  - When T and Td are far apart, RH is low
- #### Water Vapor Rule
  - Air can only hold so much water vapor
  - When it has as much water vapor as physics allow, we say the air is saturated
- ## Saturation Vapor Pressure
  - Pressure exerted by water vapor when the air is saturated with it

### Chapter 4 (cont)

- Amount of water vapor allowed in air is mostly determined by temperature
- Warmer temperatures allow for more water vapor
- ## When Air is Saturated
  - Some kind of Condensation (Deposition) Occurs
  - Dew forms
  - Frost forms (if below freezing)
  - Fog forms
  - Cloud forms
- ## Formation of Dew
  - Ideal Conditions
  - Clear skies, light winds
  - Allows for maximum radiational cooling of the ground
  - Air in contact with ground cools to the dew point
  - If air continues to cool below freezing, frozen dew occurs
- ## Formation of Frost
  - When Deposition occurs instead of condensation
- ## Clouds
  - Bringing Air to Saturation
  - Cooling air is the easiest way to saturate the air
  - Air always cools as it rises
- ### Formation of Clouds
  - Clouds result from condensation and/or deposition

### Chapter 4 (cont)

- There usually needs to be condensation nuclei for it to form
- Dust, smoke, ash, salt, sulfate particles, and even bacteria
- Without these, clouds would not form and RH would need to be greater than 100% to form
- Deposition can occur in bitterly cold air without nuclei
- Condensation or deposition will always occur when RH reaches 100% (in this class)
- ### Fog
  - Cloud with base at the ground
  - Forms when air temperature = dew temperature
  - 4 types
  - 1. Radiation
    - 1. Ground cools rapidly and causes saturation near the ground
    - 2. Nocturnal inversion can prevent higher fog
    - 3. Clear skies, light winds, high relative humidity
    - 4. Also called Valley fog
  - 2. Advection
    - 1. Warm, moist air blowing horizontally (advecting) over a "cold" surface
    - 3. Upslope
      - 1. Humid air moves up a hill or mountain



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Page 9 of 11.

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### Chapter 4 (cont)

1. The upward flow causes the air to expand, cool, which can eventually reach 100%

4. Evaporative

- Process of Evaporation involved
- Rain falls, partially evaporates
- Adds water vapor to air, leads to saturation
- Evaporation also chills air, assisting in bringing air to saturation
- Precipitation Fog
- Steam Fog
- Cool dry air moves over warm surface, esp. water
- Common over lakes in autumn when lake is still warm from summer and air above it is cold and dry

### Classification of Clouds

- Jean-Baptiste Lamarck (1802)
- Luke Howard (1803)
- Abercromby and Hildebrandsson (1887)

Clouds are classified by appearance, shape, and how high they are.

**High Clouds**

- Cirrus Clouds
- Composed mostly of ice crystals
- Thin due to limited water vapor
- Cirrocumulus
- Composed of mostly ice crystals but with lumps
- Cirrostratus

### Chapter 4 (cont)

- Composed of mostly ice crystals and when thin, often causes a halo around sun/moon
- Can be thicker but usually, sun still partially shines through

**Middle Clouds**

- Alto cumulus
- Mostly water droplets
- Darker regions noted
- Can signal the possibility of afternoon storms
- Altostratus
- Mostly water droplets with a "frosted glass" sun appearance
- Usually does not permit shadows to be cast

**Low Clouds**

- Stratus
- Layer of low clouds covering the sky
- Often seen after fog "lifts"
- Can have mist or drizzle
- Nimbostratus
- Light to moderate rain or snow
- Stratocumulus
- Lumpy clouds that appear in rows with some separation
- Lower than altocumulus with larger cloud elements

**Clouds of Vertical Development**

- Cumulus
- Cauliflower or cotton ball clouds

### Chapter 4 (cont)

- Rising air below clouds, sinking air between clouds
- Cumulus Humilis
- Fair weather clouds
- Humble clouds that do not threaten to build into storms
- Cumulus Congestus
- May develop into thunderstorms
- Cumulonimbus
- Most intense rainmaker
- Overshooting top
- When the cloud punches through the stratosphere

**Special Latin Descriptive Terms**

- Cloud Varieties
- Uncinus
- Hooked shaped, often appear before stormy weather moves in
- Cirrus uncinus
- Fractus
- Stratus or cumulus clouds that appear broken
- Cumulus fractus better known as scud clouds
- Mammatus
- Udder-shaped protuberances often associated with the underside of a cumulonimbus anvil cloud
- Only cloud that forms in sinking air

**Unusual Clouds**

- Lenticular
- Lens-shaped and common over mountains and downwind of high terrain
- Pileus
- Cap clouds

### Chapter 4 (cont)

- Form when moist air is pushed up under a developing cumulus cloud
- Banner Cloud
- Forms downwind of an isolated mountain peak
- Asperitas Clouds
- Often forming near precipitation-bearing clouds
- Undulating up and down like ocean waves

**\*Super-high Clouds\***

- Nacreous
- Stratophere
- Noctilucent
- Mesosphere

Reminder: Rising air cools, temperature drops to dew point, condenses

- Excess water vapor condenses into tiny droplets
- Excess water vapor used up very quickly
- This results in Billions of teeny tiny water droplets whose radii are 20 microns or less

**## Observing Clouds from Space**

- Polar orbiting satellite
- Geostationary satellite
- Infrared imagery provides extra detail
- Darker shades of grey indicate warm clouds, thus low altitude
- Brighter greys and whites indicate cold clouds thus high altitude

**## Water Vapor Imagery**



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Page 10 of 11.

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### Chapter 4 (cont)

- Shows air motions (wind) even in cloud-free areas
  - Detects water vapor at the 6.9 $\mu$ m wavelength
  - Colorized to differentiate between dry and moist air
- #### ## Two Main Types of Nuclei
- Condensation Nuclei
  - Hygroscopic (water-seeking) nuclei
  - Most effective condensation nuclei
  - Salt crystals are the best
  - Only RH ~75% required
  - Hydrophobic (water-repelling) nuclei
  - Least effective condensation nuclei
  - Waxes and oil droplets discourage but do not totally prevent condensation
  - RH must be 100% or even temporarily higher



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