

Chemistry - Chapter 6: Fuel Cells Cheat Sheet by Autumn (Autumn) via cheatography.com/145676/cs/31394/

Fuel Cell Vs Galvanic Cell	
Fuel Cell	Galvanic Cells
- Galvanic cell type	- Reactants contained within cell
- Reactants supplied to cell	- 60-90% efficiency
- 40-60% efficiency, up to 85%	- Portable energy
- Continuous source of high electrical currents	

Advantages Vs Disadvantages	
Advantages	Disadvantages
- More efficient (direct chemical → electrical energy)	- Expensive use/re- actants
- No CO ₂ /greenhouse gases made	- Inverter needed from DC → AC
- Electricity made on site	- Hydrogen storage and safety issues
- Many different fuels can be used (most common is hydrogen)	- Hydrogen access issues



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