

Medicinal chemistry

Definition:

Drugs: A medicine or other substances which has a marked effect when taken into body.

Medicine: A drug or other preparation is taken into body in order to treat or prevent disease

Drug Name: Adopted by company during development

Proprietary Name: Registered & used by company exclusively

Systemic Name: By IUPAC rules

Research Code: Letter & no. For company to identify the particular compound

Generic Name: Recommended international non proprietary name by WHO when patent has expired

Classifications of drug

Chemical Structure:

Common skeleton structure

May have same biological activity

Pharmacological effect:

Same biological outcome, different pathway

Limit drug use eg. :anti-fungal

Target System:

Same biological outcome different pathway(Specific chemical messengers)

Synthesis,mimics,blocks,degradation

Target Molecule:

Affects specific enzyme, neurotransmitter or receptors

Same biological outcome, same pathways

Similar structure

Membrane Permeability coefficient :

1. Depend on lipophilicity & molecular charge
2. Large coefficient = Good permeability ; Small coefficient = Poor permeability- highly charged (+/-)
3. ~80% drug posses a charge, depend on pH of the medium & pKa of a FG

pKa

Bronsted-Lowry Theory



Acid + Base \rightleftharpoons Conjugated Base + Conjugated acid

Henderson-Hasselbalch

$$pKa = pH - \log\left(\frac{[Conjugated\ base]}{[Weak\ acid]}\right)$$

pKa: pH at which 50% of substance is ionized

pKa (copy)

Bronsted-Lowry Theory



Acid + Base \rightleftharpoons Conjugated Base + Conjugated acid

Henderson-Hasselbalch



pKa (copy) (cont)

$pKa = pH - \log \left(\frac{[\text{Conjugated base}]}{[\text{Weak acid}]} \right)$

pKa: pH at which 50% of substance is ionized



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