

### SI units

mass (m)	kilograms (kg)
length (l)	metres (m)
time (t)	seconds (s)
amount of substance (n)	moles (mol)
temperature (t)	kelvin (K)
electric current (I)	amperes (A)

### Derivation of SI Units

A derived unit is comprised of a combination of SI units.

These can be derived by using the definition of the unit, or their equations eg.  $F=ma$

eg. to find the SI units of force (F), multiply the units of mass and acceleration to give  $\text{kgms}^{-2}$  (or N)

This means that every unit can be broken down into its SI base units.

### Prefixes

Tera (T)	$10^{12}$
Giga (G)	$10^9$
Mega (M)	$10^6$
Kilo (K)	$10^3$
Centi (c)	$10^{-2}$
Milli (m)	$10^{-3}$
Micro ( $\mu$ )	$10^{-6}$
Nano (n)	$10^{-9}$
Pico (p)	$10^{-12}$
Femto (f)	$10^{-15}$

These prefixes could be added before any SI units

### Conversions between units

It is possible to convert between different units of the same quantity. Here are some examples listed below:

$$1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$$

$$1 \text{ kW h} = 3\,600\,000 \text{ J or } 3.6 \text{ MJ } (\times 10^6)$$

### Types of Errors

Random error	Cause variations in both directions and are usually uncontrollable
Systematic error	Caused by faults in the experimental method or apparatus
Zero error	A type of systematic error caused by uncalibrated equipment
Parallax error	A type of systematic error caused by the apparent position of an object due to the viewing angle

### Reviewing measurements

Precise	Consistent and fluctuate around a mean value
Accuracy	A measurement that is close to the true value
Repeat-ability	The original person can redo the experiment and get the same results
Reproducibility	A different person does an experiment differently and gets the same results
Resolution	The smallest change in the quantity being measured that gives a recognisable change in reading

### Uncertainty

The bounds in which the accurate value can be expected to lie

They should be given to the same number of significant figures as the data.

### Types of Uncertainty

Absolute	Uncertainty given as a fixed quantity
Fractional	Uncertainty as a fraction of the measurement
Percentage	Uncertainty as a percentage of the measurement

### Resolution and Uncertainty

**Readings** are when one value is found

**Measurements** are when the difference between 2 readings is found

The uncertainty in a reading is  $\pm$  half the smallest division

The uncertainty in a measurement is at least  $\pm$  1 smallest division

The **resolution** of an instrument will affect its uncertainty

Digital readings and given values will either have the uncertainty quoted, or assumed to be  $\pm$  the last significant digit

For repeated data, the uncertainty is *half the range*

### Reducing Uncertainty

You can reduce uncertainty in the following ways:

- fixing one end of a ruler so there is only uncertainty in one reading
- measuring multiple times
- (for fractional and percentage) measure larger quantities

### Combining Uncertainties

Adding/subtracting data - ADD ABSOLUTE UNCERTAINTIES

Multiplying/dividing data - ADD PERCENTAGE UNCERTAINTIES

Raising to a power - MULTIPLY PERCENTAGE UNCERTAINTY BY POWER



### Uncertainties in graphs

Uncertainties are shown as error bars on graphs

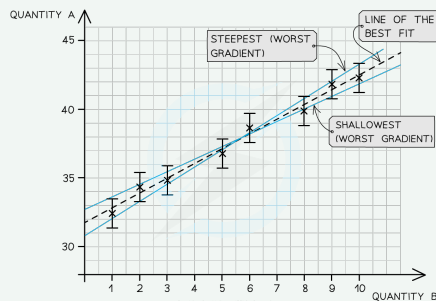
A line of best fit on a graph should go through all error bars (excluding anomalies)

The uncertainty in a gradient can be found by lines of best and worst fit

This can be done using the gradients of the steepest and shallowest lines of best fits

You can also use these two lines to find the uncertainty in the y-intercept

### Graph



### Estimation of physical quantities

*Orders of magnitude* are powers of ten which describe the size of an object

These can be used to compare the sizes of objects

Estimation is a skill used to approximate the values of physical quantities, in order to make comparisons, or to check if a value calculated is reasonable.

### Variables

Dependant	The variable that is being measured
Independent	The variable that is being changed
Control	Other variables that stay the same



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